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Tetrahedron

Tetrahedron 63 (2007) 1577-1582

Dactylospongiaquinone, a new meroterpenoid from the Australian marine sponge *Dactylospongia* n. sp.

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Received 4 October 2006; revised 20 November 2006; accepted 7 December 2006

Abstract—Chemical investigation of the sponge *Dactylospongia* n. sp. collected near Mooloolaba, S.E. Queensland, has led to the isolation of dactylospongiaquinone (7) together with the known quinones (2–5). The new metabolite 7 possesses a different carbon framework from the known dictyoceratidaquinone (9) and is suggested to possess a cis-fused ring junction by extensive NOESY studies combined with molecular modelling calculations. The relative stereochemistry of the previously described cyclospongiaquinone-1 (3) and dehydrocyclospongiaquinone-1 (4) is also assigned on the basis of NOESY analyses. Full NMR spectroscopic assignments are provided for all compounds. © 2006 Elsevier Ltd. All rights reserved.

1. Introduction

Sesquiterpene quinones are characteristic marine metabolites with examples of both drimane and 4,9-friedodrimane skeletons frequently isolated from sponge genera.¹ This class of compounds has attracted the attention of researchers because of their potent biological properties, which are reported to include antimicrobial,² antileukemic³ and immunomodulatory activities.⁴ In the late 1980s, there was significant interest in the possible anti-HIV activity of marine sesquiterpene quinones.^{1,5}

Reports on these interesting compounds from Australian sponges have been limited. In 1978, the Roche group reported the five quinones, isospongiaquinone (1), spongiaquinone (2) and the cyclic compounds cyclospongiaquinone-1 (3), dehydrocyclospongiaquinone-1 (4) and cyclospongiaquinone-2 (5), isolated from two distinct forms of *Stelospon-gia conulata* collected in New South Wales.⁶ Capon et al. subsequently assigned the absolute stereochemistry to spon-giaquinone following its isolation from a Southern Australian *Spongia* sp. along with 4 and other sesquiterpene quinones.⁷ Meanwhile, investigation of *Spongia hispida*, collected by trawling off the Victorian coast, led to the isolation of 5-*epi*-isospongiaquinone (6) with a biosynthetically interesting cis-fused ring junction.⁸ More recent investigation of an *Euryspongia* sp. has yielded novel sesquiterpene quinone/hydroquinone pairs including some chlorinated metabolites.⁹

An ongoing screening programme for bioactive sponges collected in S.E. Queensland waters identified a dictyoceratid sponge *Dactylospongia* n. sp. that showed cytotoxic and antimicrobial activities. Chemical study of this sponge has now led to the isolation of the biosynthetically novel cyclopropylsubstituted sesquiterpene quinone, named as dactylospongiaquinone (7), together with the quinones (2–5). This paper reports the structural and stereochemical studies on the isolated metabolites, together with an evaluation of their biological activities.

2. Result and discussion

Extraction of the sponge sample with DCM/MeOH 1:1 gave a dark-brown extract that was fractionated by silica gel flash chromatography (hexanes/DCM \Rightarrow DCM \Rightarrow MeOH), followed by preparative TLC using hexanes/DCM 1:4 to give spongiaquinone (2), cyclospongiaquinone-1 (3), dehydrocyclospongiaquinone-1 (4) and cyclospongiaquinone-2 (5), all identified by comparison with literature data.^{6,7} In the course of this work, one of the less polar column fractions attracted our attention since its ¹H NMR spectrum showed doublet signals at $\delta_{\rm H}$ 0.50 and -0.08 that were diagnostic of cyclopropyl ring protons. This fraction was further

Keywords: Meroterpenoid; Quinone; Cyclopropane; NMR; Sponges; Dactylospongia.

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investigated by preparative TLC and then by silica gel HPLC using hexanes/EtOAc 7:3. The HPLC isolation of the new compound **7** was complicated by the co-elution of related metabolites,⁷ hence the sample was converted to its methyl ether **8** and then subjected to further HPLC purification. Both **7** and **8** were then subjected to a detailed NMR study at 750 MHz.



The molecular formula of dactylospongiaquinone (7) was established as C₂₂H₃₀O₄ based on high-resolution mass measurement. ¹³C NMR showed diagnostic signals for a hydroxy quinone at $\delta_{\rm C}$ 181.9, 181.8, 161.0, 152.9, 117.8 and 101.9, and a methoxyl carbon at $\delta_{\rm C}$ 56.7, while the ¹H NMR spectrum showed a sole quinonoid proton at $\delta_{\rm H}$ 5.83 and a methoxyl group at $\delta_{\rm H}$ 3.83. In the sesquiterpene portion, ¹³C NMR data showed the presence of three methyls, seven methylenes, two methines and three quaternary carbons (Table 1). There were two methyl singlets at $\delta_{\rm H}$ 1.04 and 0.80, and a methyl doublet at $\delta_{\rm H}$ 0.93 (3H, d, J=6.8 Hz) linked to a multiplet at $\delta_{\rm H}$ 1.37. There were two isolated AB systems, one at $\delta_{\rm H}$ 2.73 and 2.46 (1H each, d, J=13.5 Hz; $\delta_{\rm C}$ 32.9) adjacent to the quinone ring,⁹ and the other at $\delta_{\rm H}$ 0.50 and -0.08 (1H each, d, J=4.2 Hz; $\delta_{\rm C}$ 22.4) for the cyclopropyl ring. HMBC correlations from the cyclopropyl protons identified the two other cyclopropyl carbons, at $\delta_{\rm C}$ 20.0 and 24.0. The 2D data set acquired for dactylospongiaquinone was then assessed against two carbon skeletons 7 and 9, which each contains two methyl singlets and a methyl doublet, but which differ in the positioning of the cyclopropyl group. Dictyoceratidaquinone (9) is a cyclopropyl-containing sesquiterpene quinone of undetermined stereochemistry that was isolated by Utkina and Veselova from an undescribed dictyoceratid sponge collected from the Indian Ocean.¹⁰ This group reported cyclopropyl signals at $\delta_{\rm H}$ 0.53 and 0.31, and also $\delta_{\rm C}$ 16.8 (t), 27.1 (s) and 28.0 (s)

for **9**. It was apparent from these data that the *Dactylospongia* cyclopropyl compound was not dictyoceratidaquinone. Furthermore, HMBC correlations from the methyl singlet at $\delta_{\rm H}$ 0.80 to C-11 at $\delta_{\rm C}$ 32.9 required this methyl group to be placed at C-9, consistent with structure **7**. HMBC correlations from the methyl singlet at $\delta_{\rm H}$ 1.04 to the carbons at $\delta_{\rm C}$ 20.0, 22.4 and 24.0, and from the cyclopropyl protons at $\delta_{\rm H}$ 0.50 and -0.08 to the signal at $\delta_{\rm C}$ 20.2 then placed a methyl group next to the cyclopropyl ring. TOCSY data indicated the partial structure H-6 to H-8 to Me-13. These results, together with data for the ¹H and ¹³C NMR signals of the related dimethyl ether **8** (Table 1), supported the planar structure for **7** as shown.

The relative configuration of dactylospongiaquinone was next explored. Two pieces of evidence strongly supported a cis-fused ring junction. Firstly, there was poor agreement of the ¹³C NMR data for C-3–C-10 and C-15 with the corresponding data for the cyclopropane-containing terpenes cacospongionolide,¹¹ asmarines I and J¹² and the dytesinins A and B.¹³ All these metabolites are suggested to be trans ring fused by extensive NOE studies. Secondly, in the NOESY data for methyl ether **8**, a strong correlation between the cyclopropyl proton at $\delta_{\rm H}$ 0.47 and a proton at $\delta_{\rm H}$ 1.340, assigned to H-10, could only be consistent with cis stereo-chemistry. Likewise, an NOE between H-10 ($\delta_{\rm H}$ 1.39) and a cyclopropyl proton ($\delta_{\rm H}$ 0.50) was apparent in compound **7**.

Molecular modelling linked with NOESY data provided insights into the stereochemistry at C-8 and C-9. Using the program PC-Model,¹⁴ a conformational search was carried out on each of the four possible cis-fused diastereomers (8, **10a–10c**). Figure 1 shows the lowest energy conformation for each stereoisomer, although, it should be noted that each isomer showed a number of conformations for both fused ring system and rotatable side chains. Proton-proton distances were then averaged (weighted by population) over all significant conformations. The NOEs observed between Me-12 and the individual methylene protons at C-1 of dactylospongiaquinone methyl ether were more consistent with structure 8 or 10c than with structure 10a or 10b, in which the C-9 stereochemistry is inverted. In both structures 8 and 10c, ring B adopts a chair conformation in the lowest energy conformation, and with the quinone substituent axially orientated. The observed NOEs between the H-11 protons and the H-7 signal at $\delta_{\rm H}$ 1.70 were entirely consistent with this conformational picture. The Me-13 group showed NOE correlations to the H-11 protons, and so could be placed on the same face as the quinone group. Consistent with its equatorial position, Me-13 showed NOE correlations to both H-7 protons, and there was also a strong NOE from Me-12 ($\delta_{\rm H}$ 0.80) to Me-13 ($\delta_{\rm H}$ 0.93). The NOE correlations reported in Table 1 for compound 8 all corresponded to proton-proton distances of 3.5 Å or less in one or more low energy conformations.

We then considered the relative stereochemistry of the two cyclospongiaquinone-1 metabolites **3** and **4**, since this had not been evaluated in the original Roche study.⁶ For cyclospongiaquinone-1 (**3**), the methyl group at $\delta_{\rm H}$ 0.86 assigned to H-12 showed a strong NOE to the methyl singlet at $\delta_{\rm H}$ 1.20 for H-13, therefore, these two groups were *syn*. Likewise for the dehydro compound **4**, the observation of an

Table 1. NMR assignments for compounds ${\bf 7}$ and ${\bf 8}$

| Dactylospongiaquinone 7 | | | Dactylospongiaquinone methyl ether 8 | | | |
|-------------------------|----------------------------|---|--------------------------------------|---|--------------------------------|---|
| Carbon | $\delta_{\rm C}{}^{\rm a}$ | $\delta_{ m H}{}^{ m b}$ | $\delta_{\rm C}{}^{\rm a}$ | $\delta_{ m H}{}^{ m b}$ | HMBC | Selected NOESY |
| 1 | 20.3 (t) | 1.63 (1H, m), 1.48 (1H, m) | 20.4 (t) | (a) 1.55 (1H, m), (b) 1.43 (1H, m) | H-10 | _ |
| 2 | 18.2 (t) | 1.52 (1H, m), 1.36 (1H, m) | 18.2 (t) | (b) 1.45 (111, m) (a) 1.52 (1H, m), (b) 1.20 (1H, m) | H-1, ^c H-3 | — |
| 3 | 29.3 (t) | 1.50 (1H, m), 1.42 (1H, m) | 29.0 (t) | (b) 1.59 (1H, m) β 1.52 (1H, m), α 1.40 (1H, m) | H-2 | H-14; H-15a |
| 4 | 20.0 (s) | _ | 19.8 (s) | <u> </u> | H-3, H-14, H-15 | _ |
| 5 | 24.0 (s) | _ | 23.8 (s) | _ | H-6, H-10, H-14, H-15 | _ |
| 6 | 29.2 (t) | 1.50 (1H, m), 1.16 (1H, m) | 28.4 (t) | (a) 1.50 (1H, m), (b) 1.14 (1H, m) | H-7, H-8, H-15 | H-14, H-15b; H-14, H-15b |
| 7 | 29.9 (t) | 1.67 (1H, m), 1.18 (1H, m) | 29.7 (t) | α 1.70 (1H, m), β 1.17 (1H, m) | H-6, H-8, H-13 | H-11, [°] H-13; H-8, H-13 |
| 8 | 37.0 (d) | 1.37 (1H, m) | 36.8 (d) | 1.345^{d} (1H, m) | H-7, H-11, H-12, H-13 | H-76 H-12 H-13 |
| 9 | 32.1 (s) | | 32.9 (s) | | H-8, H-10, H-11, H-12, H-13 | |
| 10 | 43.0 (d) | 1.39 (1H, m) | 43.2 (d) | 1.340 ^d (1H, m) | H-1α, H-6, H-11, H-12, H-15 | H-11, H-12, H-15a |
| 11 | 32.9 (t) | 2.73 (1H, d, <i>J</i> =13.5 Hz), 2.46 (1H, d, <i>J</i> =13.5 Hz) | 33.1 (t) | 2.76 (1H, d, <i>J</i> =13.0 Hz), 2.44 (1H, d, <i>J</i> =13.0 Hz) | H-8, H-10, H-12 | H-7α, H-10, H-12, H-13; H-7α, H-10, H-12, H-13 |
| 12 | 18.1 (q) | 0.80 (3H, s) | 18.2 (q) | 0.77 (3H, s) | H-8, H-10, H-11, H-13 | H-1, H-11, H-13 |
| 13 | 17.5 (q) | 0.93 (3H, d, J=6.8 Hz) | 17.5 (q) | 0.91 (3H, d, J=6.8 Hz) | H-7, H-8, H-11, H-12 | H-7, H-8, H-11b, H-12 |
| 14 | 20.2 (q) | 1.04 (3H, s) | 20.2 (q) | 1.04 (3H, s) | H-3, H-15 | H-3, H-6, H-15b |
| 15 | 22.4 (t) | (a) 0.50 (1H, d, $J=4.2$ Hz), (b) -0.08 (1H, d, $J=4.2$ Hz) | 22.6 (t) | (a) 0.47 (1H, d, $J=4.2$ Hz), (b) -0.06 (1H, d, $J=4.2$ Hz) | H-6, H-10, H-14 | H-3α, H-10, H-15b; H-6, H-14 H-15a |
| 1′ | 117.8 (s) | | 129.2(s) | | H-11 | _ |
| 2' | 181.8(s) | _ | 182.5(s) | _ | H-11, H-4' | _ |
| 3' | 161.0 (s) | _ | 158.9(s) | _ | H-4', 3'-OMe | _ |
| 4' | 101.9 (d) | 5.83 (1H, s) | 105.0 (d) | 5.72 (1H, s) | | 3'-OMe |
| 5' | 181.9 (s) | | 183.2 (s) | | H-4′ | |
| 6' | 152.9 (s) | _ | 157.0 (s) | _ | H-11, H-4', 6'-OMe | _ |
| 3'-OMe | 56.7 (q) | 3.83 (3H, s) | 56.7 (q) | 3.80 (3H, s) | | H-4′ |
| 6'-OH | _ | 7.28 (1H, br s) | | | _ | |
| 6'-OMe | _ | | 60.7 (q) | 4.10 (3H, s) | | |

^a Chemical shifts (ppm) referenced to CDCl₃ ($\delta_{\rm C}$ 77.0), 750 MHz. ^b Chemical shifts (ppm) referenced to CHCl₃ ($\delta_{\rm H}$ 7.25), 750 MHz. ^c Correlations to both methylene protons unless specified. ^d Distinguishable at 750 MHz.



Figure 1. Lowest energy conformation for candidate structures of dactylospongiaquinone methyl ether.

NOE from H-12 to H-13 supported the *syn* relationship. HSQC and HMBC correlations were used to assign fully the NMR data for 3-5 as presented in Section 3.

Owing to its unusual chiroptic properties, the $[\alpha]_D$ value of spongiaquinone (2) is not recorded in the literature,^{6,7} however, for (–)-spongiaquinone methyl ether (11) the absolute stereochemistry shown has been established by chemical correlation⁷ and by total synthesis.¹⁵ A sample of 11 prepared from spongiaquinone (2) isolated in this study had an $[\alpha]_D$ value of –91.3 (*c* 0.35) compared to the value of –82.2 (*c* 0.52) reported by Capon et al.⁷

Both trans- and cis-fused drimane and 4,9-friedodrimane metabolites have been isolated from marine sponges.^{1,16} A biosynthetic scheme leading to dactylospongiaquinone is shown in Figure 2, which may suggest a role for the carbocationic intermediate **12**. The stereochemistry at C-9 and C-10 results from the methyl and hydride migrations shown, while that at C-8 is of interest as it mirrors the stereochemistry found in the asmarines I and J.¹²



Figure 2. Suggested biosynthetic pathway leading to dactylospongiaquinone 7.

Spongiaquinone (2) exhibited strong cytotoxic activity against human breast cancer (BC) cells (IC₅₀ 3.24 μ g/ml), whereas cyclospongiaquinone-2 (5) showed moderate activity against human small cell lung cancer (NCI-H187) cells (IC₅₀ 4.96 μ g/ml). Both compounds were inactive in anti-tubercular and antimalarial screens.

3. Experimental

3.1. General experimental procedures

Optical rotations were obtained using a JASCO-P1010 polarimeter. 1D and 2D NMR spectra were acquired using Bruker DRX-500 or Bruker DMX-750 instrument. NMR spectra were obtained in deuterochloroform at room temperature. Samples were internally referenced to CHCl₃ at $\delta_{\rm H}$ 7.25 or CDCl₃ at $\delta_{\rm C}$ 77.0. High- and low-resolution mass measurements were obtained from a Finnigan MAT 900 XL-Trap electrospray (ESI) mass spectrometer with a Finnigan API III electrospray source.

3.2. Biological material

Specimens of *Dactylospongia* n. sp. were collected from the Inner Gneerings, a group of shoals near Mooloolaba (Australia), using SCUBA at a depth of 10–15 m on 16 January 2006. Samples were taken back to the laboratory where they were stored at -20 °C until extraction. The sponge was charcoal grey on the surface and orange-yellow on the underside. The shape was globular and the sample was approximately 10 cm thick. A voucher specimen (QM G324323) is lodged at the Queensland Museum. Photographs of the sponge material are available from the authors.

3.3. Extraction and isolation of quinones

The specimen of *Dactylospongia* n. sp. (wet weight 163 g) was cut into small pieces and extracted exhaustively with MeOH. The extract was removed, filtered through cotton and then evaporated under reduced pressure to give an aqueous residue, which was partitioned sequentially with hexanes, DCM, EtOAc and finally n-BuOH. The hexane fraction was dried over anhydrous Na2SO4 and concentrated under reduced pressure to give 960 mg of a brown solid, which was analysed by TLC and ¹H NMR. The extract was subjected to gradient elution Si flash chromatography (hexanes \Rightarrow DCM \Rightarrow MeOH). The fractions that eluted in hexanes/DCM (1:3) were combined and analysed by TLC and ¹H NMR. The EtOAc/hexanes (3:7) soluble portion of this fraction was purified using semi-preparative NP-HPLC (Waters 515; Gilson 132 series RI detector; Waters $10\mu \mu Porasil 7.8 \times 300 \text{ mm}$ column; flow rate 2.2 ml/min) with EtOAc/hexanes (3:7) as solvent to afford 7. The fractions that eluted in hexanes/DCM (1:4) were combined and analysed by TLC and ¹H NMR yielding spongiaquinone (2), cyclospongiaquinone-1 (3), dehydrocyclospongiaquinone-1 (4) and cyclospongiaquinone-2 (5).

3.3.1. Spongiaquinone (2).^{6,7} Compound 2 (1.20 mg) was obtained as a red amorphous solid: $[\alpha]_D$ -50.0 (c 0.07, CHCl₃); ¹H NMR (CDCl₃, 500 MHz) $\delta_{\rm H}$ 5.85 (1H, s, H-4'), 5.53 (1H, s, H-11), 3.83 (3H, s, OMe), 2.52 (1H, m, H-8), 1.80 and 1.49 (1H each, m, H-1), 1.70 and 1.45 (1H each, m, H-2), 1.55 (2H, m, H-6), 1.54 (2H, m, H-7), 1.37 and 1.22 (1H each, m, H-3), 1.19 (3H, s, H-13), 1.09 (3H, d, J=7.0 Hz, H-12), 0.87 (3H, s, H-14) and 0.87 (3H, s, H-15); ¹³C NMR (CDCl₃, 500 MHz) $\delta_{\rm C}$ 181.5 (C, C-5'), 181.0 (C, C-2'), 162.9 (C, C-9), 161.1 (C, C-3'), 150.0 (C, C-6'), 117.0 (C, C-1'), 106.8 (CH, C-11), 102.1 (CH, C-4'), 56.5 (OMe-3'), 54.3 (C, C-5), 41.9 (CH₂, C-3), 41.1 (C, C-10), 38.3 (CH₂, C-1), 33.9 (C, C-4), 33.8 (CH₃, C-15), 33.7 (CH₂, C-7), 33.1 (CH, C-8), 22.3 (CH₃, C-13), 21.6 (CH₃, C-12), 21.5 (CH₃, C-14), 18.8 (CH₂, C-2) and 17.6 (CH₂, C-6); LRESIMS *m*/*z* 381 (M+Na)⁺.

3.3.2. Cyclospongiaquinone-1 (3).⁶ Compound 3 (1.30 mg) was obtained as a yellow amorphous solid: $[\alpha]_D +94.6$ (*c* 0.06, CHCl₃), lit.⁶ $[\alpha]_D -2.15$ (*c* 1, CHCl₃); ¹H NMR (CDCl₃, 500 MHz) δ_H 5.73 (1H, s, H-4'), 3.78 (3H, s, OMe), 2.49 and 2.06 (1H each, m, H-11), 2.20 and 1.76 (1H each, m, H-1), 1.75 and 1.31 (1H each, m, H-6), 1.72 and 0.75 (1H each, m, H-7), 1.60 and 1.45 (1H each, m, H-2), 1.42 (1H, dd, *J*=18.0, 4.0 Hz, H-9), 1.20 (3H, s, H-13), 0.97 (1H, m, H-5), 0.87 (3H, s, H-14), 0.86 (3H, s, s).

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H-12) and 0.81 (3H, s, H-15); ¹³C NMR (CDCl₃, 500 MHz) $\delta_{\rm C}$ 181.2 (C, C-5'), 181.0 (C, C-2'), 161.1 (C, C-3'), 152.0 (C, C-6'), 116.0 (C, C-1'), 104.5 (CH, C-4'), 81.7 (C, C-8), 56.2 (OMe-3'), 55.9 (CH, C-5), 51.2 (CH, C-9), 41.6 (CH₂, C-3), 40.1 (CH₂, C-1), 39.1 (CH₂, C-7), 37.0 (C, C-10), 33.3 (CH₃, C-15), 33.0 (C, C-4), 21.4 (CH₃, C-14), 20.3 (CH₃, C-13), 19.6 (CH₂, C-6), 18.4 (CH₂, C-2), 16.1 (CH₂, C-11) and 14.9 (CH₃, C-12); LRESIMS *m*/*z* 381 (M+Na)⁺.

3.3.3. Dehydrocyclospongiaquinone-1 (4).⁶ Compound 4 (0.40 mg) was obtained as an orange amorphous solid: sample decomposed before an $[\alpha]_D$ value could be measured; ¹H NMR (CDCl₃, 500 MHz) δ_H 6.24 (1H, s, H-11), 5.73 (1H, H-4'), 3.79 (3H, s, OMe), 2.33 and 1.94 (1H each, m, H-7), 1.98 and 1.35 (1H each, m, H-1), 1.64 (2H, m, H-6), 1.60 and 1.14 (1H each, m, H-2), 1.49 (3H, s, H-13), 1.40 and 1.12 (1H each, m, H-3), 1.13 (3H, s, H-12), 1.03 (1H, m, H-5), 0.89 (3H, s, H-14) and 0.84 (3H, s, H-15); ¹³C NMR (CDCl₃, 500 MHz) δ_C 180.0 (C, C-5'), 179.0 (C, C-2'), 158.8 (C, C-3'), 151.6 (C, C-9), 149.4 (C, C-6'), 118.0 (C, C-1'), 106.8 (CH, C-11), 104.9 (CH, C-4'), 82.8 (C, C-8), 56.3 (OMe-3'), 51.6 (CH, C-5), 41.1 (CH₂, C-3), 41.0 (CH₂, C-7), 39.7 (C, C-10), 37.8 (CH₂, C-1), 33.4 (C, C-4), 33.0 (CH₃, C-15), 26.8 (CH₃, C-13), 23.4 (CH₃, C-12), 21.6 (CH₃, C-14), 21.5 (CH₂, C-6), 18.8 (CH₂, C-6) and 18.6 (CH₂, C-2); LRESIMS *m*/*z* 379 (M+Na)⁺.

3.3.4. Cyclospongiaquinone-2 (5).⁶ Compound 5 (1.3 mg) was obtained as an orange amorphous solid: $[\alpha]_D - 18.1$ (c 0.15, CHCl₃), lit.⁶ $[\alpha]_D$ +11.68 (c 1, CHCl₃); ¹H NMR (CDCl₃, 500 MHz) $\delta_{\rm H}$ 5.67 (1H, s, H-4'), 3.83 (3H, s, OMe), 3.38 and 2.64 (1H each, d, J=17.0 Hz, H-11), 2.06 (1H, m, H-8), 1.62 (2H, m, H-7), 1.61 and 1.50 (1H each, m, H-2), 1.46 (2H, m, H-6), 1.35 and 1.10 (1H each, m, H-1), 1.25 and 1.14 (1H each, m, H-3), 1.18 (3H, s, H-12), 1.18 (3H, d, J=7.0 Hz, H-13), 0.97 (1H, m, H-5), 0.89 (3H, s, H-15) and 0.87 (3H, s, H-14); ¹³C NMR (CDCl₃, 500 MHz) δ_C 179.0 (C, C-2'), 178.6 (C, C-5'), 160.8 (C, C-3'), 159.2 (C, C-6'), 118.1 (C, C-1'), 104.1 (CH, C-4'), 103.1 (C, C-9), 56.7 (OMe-3'), 48.7 (CH, C-5), 43.3 (C, C-10), 41.6 (CH₂, C-1), 40.8 (CH, C-8), 40.5 (CH₃, C-13), 36.8 (CH₂, C-11), 33.2 (CH₃, C-15), 32.9 (C, C-4), 31.8 (CH₂, C-3), 29.1 (CH₂, C-7), 21.1 (CH₃, C-14), 16.8 (CH₂, C-2), 16.6 (CH₂, C-6) and 16.6 (CH₃, C-12); LRESIMS m/z 381 (M+Na)⁺.

3.3.5. Dactylospongiaquinone (7). Compound **7** (0.58 mg) was obtained as a yellow amorphous solid: $[\alpha]_D - 27.9$ (*c* 0.03, CHCl₃); ¹H and ¹³C NMR (CDCl₃, 750 MHz) see Table 1; HRESIMS *m*/*z* 381.2036, calcd for C₂₂H₃₀O₄Na 381.2042.

3.3.6. Dactylospongiaquinone methyl ether (8). A sample of 7 (1.8 mg, 0.005 mmol) was stirred with MeI (0.47 µl, 0.0075 mmol, 1.5 equiv) and K₂CO₃ (0.76 mg, 0.0055 mmol, 1.1 equiv) in anhydrous DMF (20 µl) at room temperature overnight. The reaction mixture was diluted with H₂O and extracted with CH₂Cl₂ (3×5 ml). The organic layer was washed with H₂O and dried (Na₂SO₄). After filtration through a cotton wool plug and removal of the solvent, the crude methyl ether was further purified by semi-preparative NP-HPLC (20% EtOAc/hexanes, 1.5 ml/min (2:8)) to afford 8 (0.32 mg, 17%) as a yellow amorphous solid: $[\alpha]_{D}^{22} - 16.4$

(*c* 0.05, CHCl₃); for ¹H and ¹³C NMR (CDCl₃, 750 MHz) data see Table 1; HRESIMS m/z 395.2205, calcd for C₂₃H₃₂O₄Na 395.2198.

3.3.7. Spongiaquinone methyl ether (11). Compound **11** was prepared similarly as a yellow oil: $[\alpha]_D -91.3$ (*c* 0.35, CHCl₃), lit.⁷ $[\alpha]_D -82.2$ (*c* 0.52, CHCl₃); NMR data were in accordance with literature values.⁷

3.4. Cytotoxicity assays

The cytotoxicity assays against BC and NCI-H187 cells were performed employing a colourimetric method.¹⁷ The standard drug ellipticine exhibited IC_{50} values against these cell lines at 1.46 and 0.39 µg/ml, respectively.

Acknowledgements

We thank the Royal Golden Jubilee Fund, Thailand for a Ph.D. scholarship (to A.J.), the Australia Research Council for funding, L. Lambert (Centre for Magnetic Resonance, UQ), G. MacFarlane of the School of Molecular and Microbial Science, UQ, for spectroscopic assistance, Professor R. Capon, Institute for Molecular Bioscience, UQ, for assistance with the measurements of $[\alpha]_D$ values, and K. Rands-Trevor for her assistance with the preparation of compound **12**. ScubaWorld Mooloolaba assisted with access to dive sites for the sample collection (carried out under permit from the Department of Primary Industries and Fisheries, Queensland).

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